

2 3 Elements And Compounds

2.3 Elements and Compounds. You can use their general characteristics to distinguish substances from mixtures. If the composition of a material is fixed, the material is a substance. If the composition of a material may vary, the material is a mixture.

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Elements and Compounds. Take two pounds of sugar, two cups of boiling water, and one quarter teaspoon of cream of tartar. Add food coloring and you have the sticky, sweet concoction known as cotton candy. You will learn how substances are classified as elements or compounds.

Chemistry 2.3 Elements and Compounds

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Figure 2.10 Compounds and the elements from which they form have different properties. Observing Based on the photographs, describe two physical properties of sodium and two of chlorine. (commonly known as table salt) is used to season or preserve food. Chlorine is used to kill harmful organisms in swimming pools.

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Compounds are made up of , which are always present in 2. the same in a given compound. Compounds can be 3. separated into their constituent elements only by reaction. 4. Each element is represented by a one or two letter . 5. For example, carbon is represented by the symbol , while 6. potassium is represented by the symbol . 7. 8. Part B True-False

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Compounds A compound is a substance that contains atoms of two or more different elements, and these atoms are chemically joined together. For example, water is a compound of hydrogen and oxygen.

Atoms, elements and compounds - Revision 2 - KS3 Chemistry ...

There are over 100 different elements, which are made up of atoms. Elements can be divided into metals and non-metals. Chemical symbols and formulae are used to represent elements and compounds.

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